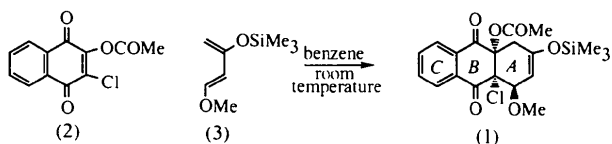
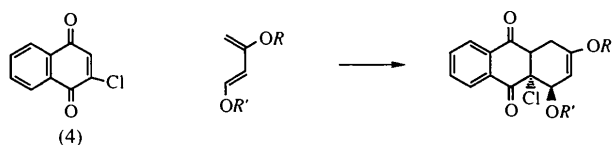


- Evans, P. A., Holmes, A. B. & Russell, K. (1994). *J. Chem. Soc. Perkin Trans. 1*, pp. 3397–3409.
- Fujii, N., Habashita, H., Akaji, M., Nakai, K., Ibuka, T., Fujiwara, M., Tamamura, H. & Yamamoto, Y. (1996). *J. Chem. Soc. Perkin Trans. 1*, pp. 865–866.
- Hammond, G. S. (1955). *J. Am. Chem. Soc.* **77**, 334–338.
- Handbook of Chemistry & Physics (1993–94). Edited by D. R. Lide, pp. 9-1 to 9-41. Boca Raton: CRC Press.
- Kumar, A. & Dittmer, D. C. (1994). *J. Org. Chem.* **59**, 4760–4764.
- Leffler, J. E. (1953). *Science*, **117**, 340–341.
- Molecular Structure Corporation (1988). *MSC/AFC Diffractometer Control Software*. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
- Molecular Structure Corporation (1995). *TEXSAN. Single Crystal Structure Analysis Software*. Version 1.7. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.
- Pauling, L. (1947). *J. Am. Chem. Soc.* **69**, 542–553.
- Pauling, L. (1960). *The Nature of the Chemical Bond*, p. 239. Ithaca: Cornell University Press.

orientation with respect to the chloro group as for additions involving monochloro quinones, e.g. compound (4) (see Scheme II) (Cameron, Feutrill & Keep, 1989).



Scheme I



Scheme II

Compound (1) is a linear tricycle with a *cis* junction between rings A and B. Both rings A and B exist in slightly distorted half-chair conformations with approximate local axes of symmetry bisecting the C1—C6 and C3—C4 bonds for ring A, and the C1—C6 and C8—C13 bonds for ring B. The methoxy substituent occupies a pseudo-axial position [O6—C5—C4—C3 – 98.7 (5)°] and is antiperiplanar to the chloro substituent [O6—C5—C6—Cl – 163.6 (3)°]. The chloro substituent is pseudo-equatorial with respect to ring B and pseudo-axial with respect to ring A, while the acetoxy substituent is pseudo-axial with respect to ring B and pseudo-equatorial with respect to ring A.

Acta Cryst. (1997). **C53**, 386–387

(1 α ,4 α β ,9 α β)-4a-Acetoxy-9a-chloro-1-methoxy-3-trimethylsilyloxy-1,4,4a,9,9a,10-hexahydroanthracene-9,10-dione

JONATHAN M. WHITE, ANDREW G. RICHES, DONALD W. CAMERON AND PETER G. GRIFFITHS

School of Chemistry, University of Melbourne, Parkville, Victoria 3052, Australia. E-mail: jonathan.white@muwayf.unimelb.edu.au

(Received 2 August 1996; accepted 31 October 1996)

Abstract

The structure of the title compound, (1 α ,4 α β ,9 α β)-9a-chloro-1-methoxy-9,10-dioxo-3-trimethylsilyloxy-1,4,4a,9,9a,10-hexahydro-4a-anthracenyl acetate, C₂₀H₂₃ClO₆Si, was determined in order to ascertain the regiochemistry of the Diels–Alder reaction between acetoxychloronaphthoquinone and a 1,3-dioxybutadiene.

Comment

The title compound, (1), was the sole product obtained from the Diels–Alder cycloaddition reaction between acetoxychloronaphthoquinone (2) (Fries & Ochwat, 1923) and 1-methoxy-3-(trimethylsilyloxy)butadiene, (3) (Scheme I). The present structure analysis establishes that the regiochemistry of cycloaddition is such that the nucleophilic methylene terminus of the diene has attacked *ipso* to the acetoxy group; this is the same

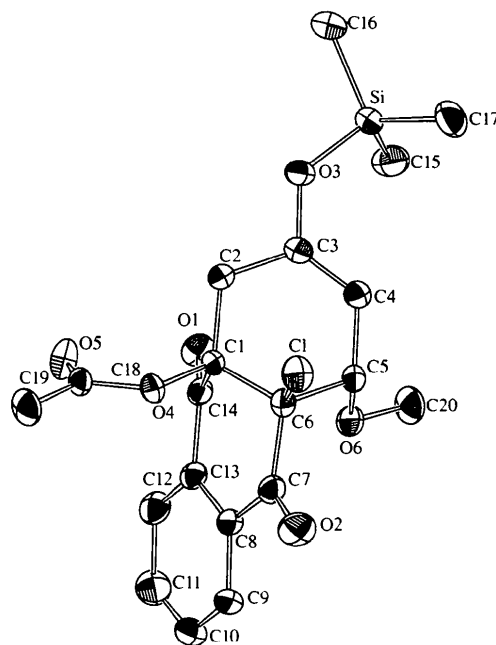


Fig. 1. ZORTEP (Zsolnai, 1994) diagram of (1). Displacement ellipsoids are plotted at the 30% probability level.

Experimental

Colourless rods of (1) (m.p. 397–402 K) were grown from petrol.

Crystal data

C₂₀H₂₃ClO₆Si

M_r = 422.94

Monoclinic

*P*2₁

a = 8.137 (2) Å

b = 12.363 (2) Å

c = 10.537 (2) Å

β = 98.77 (2)°

V = 1047.6 (4) Å³

Z = 2

D_x = 1.341 Mg m⁻³

D_m not measured

Cu *K*α (Ni-filtered) radiation

λ = 1.5418 Å

Cell parameters from 25 reflections

θ = 25–30°

μ = 2.453 mm⁻¹

T = 293 (2) K

Rod

0.40 × 0.20 × 0.15 mm

Colourless

Data collection

Enraf–Nonius CAD-4S diffractometer

ω/2θ scans

Absorption correction:

Gaussian (SHELX76);

Sheldrick, 1976)

T_{min} = 0.50, *T_{max}* = 0.72

1957 measured reflections

1832 independent reflections

1726 reflections with

I > 2σ(*I*)

R_{int} = 0.0511

θ_{max} = 63.96°

h = 0 → 9

k = 0 → 14

l = -12 → 12

3 standard reflections

frequency: 160 min

intensity decay: <2%

Refinement

Refinement on *F*²

R(*F*) = 0.0397

wR(*F*²) = 0.1046

S = 1.235

1832 reflections

346 parameters

H atoms: see below

w = 1/[σ²(*F_o*²) + (0.0505*P*)² + 0.4093*P*]

where *P* = (*F_o*² + 2*F_c*²)/3

(Δ/σ)_{max} = 1.010

Δρ_{max} = 0.238 e Å⁻³

Δρ_{min} = -0.354 e Å⁻³

Extinction correction:

SHELXL93

Extinction coefficient:

0.0073 (10)

Scattering factors from

International Tables for Crystallography (Vol. C)

Absolute configuration:

Flack (1983)

Flack parameter = 0.01 (3)

The title structure was solved by direct methods using SHELXS86 (Sheldrick, 1990). Refinement was performed with SHELXL93 (Sheldrick, 1993) using anisotropic displacement parameters for all non-H atoms and isotropic displacement parameters for the H atoms. All calculations were carried out on a VAXstation 4000VLC computer system.

Data collection: CAD-4/VAX (Enraf–Nonius, 1989). Cell refinement: CAD-4/VAX. Molecular graphics: ZORTEP (Zsolnai, 1994). Software used to prepare material for publication: SHELXL93.

The authors acknowledge support from the Australian Research Grants Scheme and the Australian Postgraduate Award (to AGR).

Lists of atomic coordinates, displacement parameters, structure factors and complete geometry have been deposited with the IUCr (Reference: OA1012). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

References

- Cameron, D. W., Feutrill, G. I. & Keep, P. L. C. (1989). *Tetrahedron Lett.* **38**, 5173–5176.
- Enraf–Nonius (1989). *CAD-4/VAX Software*. Enraf–Nonius, Delft, The Netherlands.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Fries, K. & Ochwat, P. (1923). *Ber. Dtsch. Chem. Ges.* **56**, 1291–1304.
- Sheldrick, G. M. (1976). SHELX76. *Program for Crystal Structure Determination*. University of Cambridge, England.
- Sheldrick, G. M. (1990). *Acta Cryst.* **A46**, 467–473.
- Sheldrick, G. M. (1993). SHELXL93. *Program for the Refinement of Crystal Structures*. University of Göttingen, Germany.
- Zsolnai, L. (1994). ZORTEP. *An Interactive ORTEP Program*. University of Heidelberg, Germany.

Acta Cryst. (1997). **C53**, 387–391

(1*R*,1*R'*)-2-*exo*-Mercapto-2'-thio-3-*exo*,3'-*exo*-bibornane, 2-Dehydro-2,2'-*exo*-epidithio-3,3'-bibornane and 2-*endo*,2'-*exo*-Epidithio-3,3'-bibornanylidene. Potential Antiviral Agents

WITOLD KWIATKOWSKI,^a T. S. CAMERON,^a PAUL SALAMA^b AND MARC POIRIER^b

^aDepartment of Chemistry, Dalhousie University, Halifax, Nova Scotia, Canada B3H 4J3, and ^bDépartement de Chimie, Université de Moncton, Moncton, New Brunswick, Canada E1A 3E9. E-mail: witek@scameron-10.chem.dal.ca

(Received 8 May 1996; accepted 30 September 1996)

Abstract

The compounds 2,2'-*exo*-epidithio-1,1',7,7',7'-hexamethyl-3,3'-bibicyclo[2.2.1]hept-2-ene, C₂₀H₃₀S₂, (2), and 2-*endo*,2'-*exo*-epidithio-1,1',7,7',7'-hexamethyl-3,3'-bibicyclo[2.2.1]heptanylidene, C₂₀H₃₀S₂, (3), were prepared as potential antiviral agents and their structures were determined by X-ray diffraction. It has been shown previously that there is a relationship between the strain energy of a C—S—S—C group, as measured by its planarity and S—S bond length, and the antiviral activity of the compound containing this group. In